Density Functional Theory III

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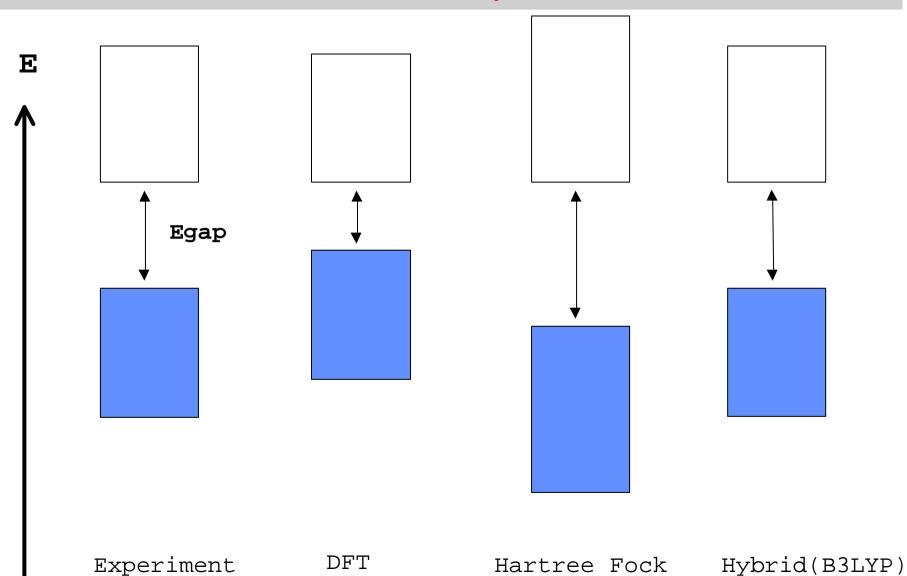
Conclusions II

- Approximations to DFT are currently the state of the art for materials simulation but do not provide a systematic approach to the exact result.
- With a judicious choice of functional atomisation energies are *typically* accurate to 3-5 kcal/mol, structures to 0.01 Ang., frequencies to 40-60 cm⁻¹
- Much larger errors are possible in 'difficult' systems
- Heavily parameterised functionals gain a little for the training set but appear to be less transferable
- Hybrid and meta-GGA look very promising.

Band Gaps!

- The band gap determines many properties: optical, dielectric, electronic....
- Formally DFT yields the total energy and density the eigenvalues are **NOT** excitation energies
- The DFT eigenvalues do have some correspondence with the observed adsorbtion spectra...

DFT Band Structures – Qualitative Guide!



Self Interaction.. I

The potential in DFT is computed from the TOTAL density.

$$\left[-\frac{1}{2} \nabla^2 + v_{ext}(\mathbf{r}) + \int \frac{\mathbf{r}(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d\mathbf{r}' + V_{xc}(\mathbf{r}) \right] \mathbf{f}_i(\mathbf{r}) = \mathbf{e}_i \mathbf{f}_i(\mathbf{r})$$

$$V_{xc}(\mathbf{r}) = \frac{\partial E_{xc}[\mathbf{r}]}{\partial \mathbf{r}(\mathbf{r})} \qquad \mathbf{r}(\mathbf{r}) = \sum_{i=1}^{N} |\mathbf{f}_{i}(\mathbf{r})|^{2}$$

As each individual electron contributes to this density – an electron interacts with itself!

Or...

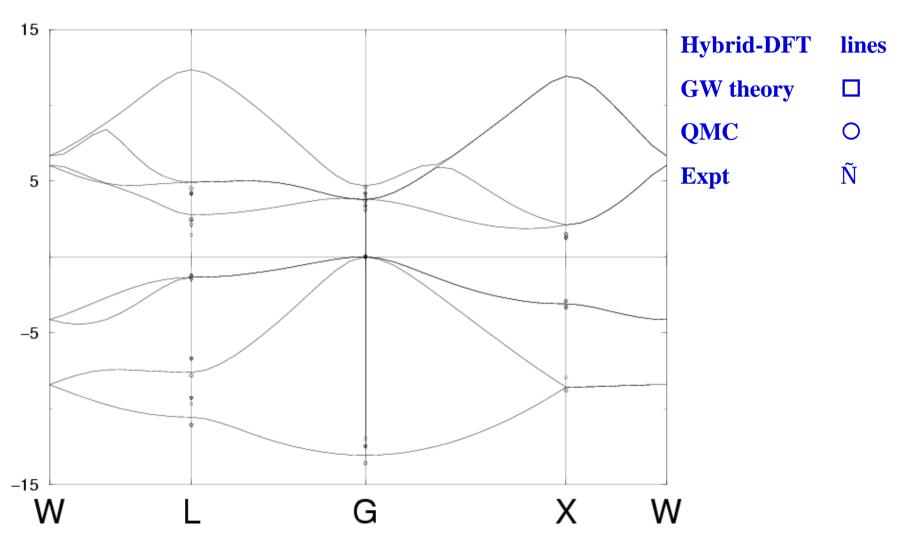
Self Interaction II

- An electron in an occupied orbital interacts with N electrons V(N) when it should interact with N-1 ie: V(N-1) this is the *self interaction*.
- DFT: All occupied bands are pushed up in energy by this interaction essentially an on-site/diagonal Coulomb/Exchange repulsion. DFT band gaps are TOO SMALL..
- Hartree Fock theory... Exchange and Coulomb cancel exactly no self interaction. In fact over corrected GAPS TOO BIG.
- Hybrid 20% HF and 80% DFT Gaps are pretty good !!!

Hybrid Functionals and Band Gaps

	Expt (eV)	Hybrid (eV)
Si	~3.5	3.8
Diamond	5.5	5.8
GaAs	1.4	1.5
ZnO	3.4	3.2
Al_2O_3	~9.0	8.5
Cr ₂ O ₃	3.3	3.4
MgO	7.8	7.3
MnO	3.6	3.8
NiO	3.8	3.9
TiO ₂	3.0	3.4
FeS ₂	1.0	2.0
ZnS	3.7	3.5

The Band Structure of Silicon



J. Muscat, A. Wander, N.M. Harrison Chem. Phys. Lett 2001.